

High School Definition of a Fluid

- Three most common phases of matter are solid, liquid and gas.
- Liquids and gases are together called fluids.

	Intermolecular Attraction Forces	Molecules	Volume and Space
Solid	Strong	Relative positions are rather fixed	Definite volume
			Definite shape
Liquid	Medium	Free to change their relative positions	Definite volume
			Indefinite shape
Gas	Weak	Practically unrestricted	Indefinite volume
			Indefinite shape

- Later we'll give another definition for "fluid", based on its behavior under shear forces.
- Exercise : From a technical point of view, what is the difference between gases and vapors?
- Exercise : What are the fourth and fifth phases of matter? Do a research on them.

Mechanics

- Mechanics studies the motion and deformation of material bodies under applied loads (forces, moments).
- It involves loads, energy, motion, deformation and material properties.
- When the material is in solid phase it is called solid mechanics, which you already studied in ME 205, ME 206, ME 208.
- When the material is in liquid or gas phase it is called fluid mechanics.



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Branches of Science that Study Fluids

- Mechanics is concerned with both stationary and moving bodies under the influence of loads.
 - Statics is the branch of mechanics that deals with bodies at rest.
 - Dynamics is the branch of mechanics that deals with bodies in motion.
- Fluid mechanics deals with the behavior of fluids at rest (fluid statics) and in motion (fluid dynamics).
 - Hydrodynamics studies liquids (incompressible flow) in motion.
 - Hydraulics studies liquids flowing in pipes, ducts and open channels.
 - Gas dynamics studies compressible flow of gases with high density changes.
 - Aerodynamics is similar to gas dynamics, but also covers low speed flows. It focuses on air flow.

















Significant Figures

Exercise: In another midterm exam of ME 305 a different version of the previous question was asked. This time numerical values are given for all the parameters, including the viscosity of the fluid, and the required torque is asked. One student provided the following answer and the instructor put a comment on the student's paper saying "Are you serious?". What does the instructor expect from the student?

Student's answer: Torque = 15.908633 Nm

- Number of significant figures that need to be used in an answer depends on the significant figures of the numbers that you used to evaluate it.
- For measured quantities, number of significant figures depends on the minimum reading scale of the measuring device.



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• Often an engineer needs to use his/her engineering intuition to judge the correct number of significant figures to use.

Fundamental Flow and Fluid Properties

Density	ρ	kg/m³
Velocity	\vec{V}	m/s
Pressure	р	Pa , atm , bar , mmHg
Viscosity	μ	Pa·s, poise
Temperature	Т	K or ℃
Internal energy	ŭ	J/kg
Enthaply	h	J/kg
Entropy	S	J/(kg K)
Specific heat	C_p , C_v	J/(kg K)
Thermal conductivity	k	W/(m K)

It is common to use p and T to fix the thermodynamic state. Then other properties can be expressed as a function of these two

$$\rho = \rho(p,T)$$
, $h = h(p,T)$, $\mu = \mu(p,T)$, etc.

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Density

- (ρ) [kg/m³]
- Mass contained in a unit volume of a fluid. $\rho = m/\forall$
- Density determines the inertia of a unit volume of fluid and hence its acceleration when subjected to a given force. Gases are easier to accelerate than liquids.
- Density also determines the amount of gravitational force (weight) acting on a fluid body. Weight of gases are neglected more often than that of liquids.
- Fluids have a very wide range of density.

	Hydrogen Gas	Methane (Natural Gas)	Air	Water	Mercury
Density [kg/m³]	0.084	0.67	1.2	998	13600
(at standart conditions)					

Density Density in general is a function of *p* and *T*, i.e. *ρ* = *ρ* (*p*,*T*) If a fluid's density is a function of pressure only (not temperature) it is called a barotropic fluid, a simplification mostly used in meteorology. Following processes demonstrate how density of an ideal gas (*p* = *ρRT*) can change with temperature and pressure.













Velocity Field

Fluid velocity: (\vec{V}) [m/s]

• In different coordinate systems velocity vector components are

Cartesian : $\vec{V} = u\vec{i} + v\vec{j} + w\vec{k}$ Cylindrical : $\vec{V} = V_r\vec{e_r} + V_\theta\vec{e_\theta} + V_z\vec{e_z}$

 In general velocity field of a flowing fluid is too complicated to be expressed as a closed form equation of space and time.





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Visualization of flow over an airfoil Visualization of flow over an http://www.ecourses.ou.edu

























	Viscosity (cont'd)					
 Definition of fluid : A fluid deforms <u>continuously</u> under the application of a shear (tangential) force, <u>no matter how small</u> the force is. 						
•	It is more difficult to deform highly viscous fluids.					
	Air	Water	SAE 30 oil	Glycerin	Thick Molasses	
	μ_{air}	50 μ_{air}	15,000 μ_{air}	75,000 μ_{air}	375,000 μ _{air}	
•	 Viscosity can be measured using capillary tube viscometer falling sphere viscometer concentric cylinder viscometer Saybolt viscometer that you'll be using in the first experiment of ME 305 					
•	Kinematic vise	cosity:(v) [m [:]	ν^2/s] $\nu = \mu$	Greek	etter "nu"	1





Viscous Behavior of Fluids (cont'd) Newtonian behavior simple (it is linear). Common fluids such as water, air, oils behave as Newtonian. Inviscid (ideal) fluids have μ = 0 and they do not exist in real world. It is an idealization. Bingham plastics do not flow below a certain amount of shear stress. (toothpaste, mayonnaise). Shear thinning fluid become thinner under increased shear stress. (wall paint, blood). Shear thickening fluids become thicker under increased shear stress. (printing ink, com starch-water mixture, quicksand). For thixotropic fluids viscosity decreases with time (the longer the shear force is applied) (lipstick). For rheopectic fluids viscosity increases with time (solidifying concrete).

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